# Hypervalent Iodine Reagents: Versatile Reagents in Synthetic Chemistry

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#### ABSTRACT

The plethora of reactions that may be carried out with organohypervalent iodine reagents in outstanding yield and selectivity has made them more desirable and adaptable organic synthesis reagents in recent years. Hypervalent iodine reagents in comparison with toxic heavy metal reagents are milder and having similar reactivity. Heterocyclic compounds are biologically active molecules and are generally used in the formation of, agrochemicals, drugs, polymeric materials and dyes. Organic chemists have been engaged in developing a number of newer reagents to synthesize such kind of heterocyclic compounds. Due to their low toxicity, simple handling, and generally benign nature, organohypervalent iodine compounds have garnered a great deal of interest for use in organic synthesis. **Keywords:** Organohypervalent iodine reagents, heterocyclic compounds, iodobenzene diacetate, iodosylbenzene, dichloroiodobenzene, [hydroxy(tosyloxy)iodo] benzene.

#### 1. INTRODUCTION

These compounds are referred to as "hypervalent" because they contain donor atoms that have more valences than would be predicted by conventional theory. The strong electrophilic character of the iodine combined with the phenyliodonio group's tendency to leave groups make these chemicals more reactive to nucleophilic attack.

Our and other research groups' numerous beneficial oxidative changes using iodine(III) have demonstrated that [hydroxy(tosyloxy)iodo] benzene (HTIB; Koser's reagent), dichloroiodobenzene (PhICl<sub>2</sub>), iodobenzene diacetate (IBD), iodosylbenzene (PhIO)<sub>n</sub>, 2-iodoxy-benzoic acid (IBX), Dess-Martin Periodinane (DMP) etc. find stimulating applications in the synthesis of heterocyclic compounds.

The results of a vast amount of study have demonstrated that the oxidizing nature of organic iodine reagents is their most crucial characteristic. In the synthesis of numerous selective oxidative transformations of many complex chemical molecules, hypervalent iodine reagents are now often utilized. Organohypervaleniodine reagents have grown significantly over the past several decades in synthetic organic chemistry. Despite the fact that the topic has been extensively covered in the literature through numerous review articles <sup>[1-6]</sup> and books <sup>[7]</sup>, there is a rising need to investigate fresh uses for these reagents. These chemicals have been used to carry out a wide range of chemical transformations, including oxidative cyclizations, oxidative coupling, oxidative rearrangements, and  $\alpha$ -functionalization of carbonyl compounds. Iodine(III)-mediated oxidative approach from several sources results in several beneficial changes.

Here, synthetic utility of hypervalent iodine reagents particularly, HTIB and iodosobenzene (IB) has been described.



Heterocyclic compounds are considered one of the most important and extensively studied divisions of organic chemistry. Theses play a crucial role in medicinal chemistry and are of great interest to researchers in the pharmaceutical industry. The increasing research interest in heterocycles can be attributed to their wide range of properties and potential applications. <sup>[8, 9]</sup>

This chapter has been structured in various segments according to the types of the heterocyclic compounds.

## 2. SYNTHESIS OF HETEROCYCLIC COMPOUNDS

## A. Sythesis of oxazoles

Nachtsheim and Hempel developed an effective way to make oxazoles through oxidative cyclization by treating Nstyrylbenzamides with [bis(trifluoroacetoxy)iodo]benzene and TMSOTf (**Scheme 1**). <sup>[10]</sup> Similar to this, Boto and colleagues reported on the one-pot synthesis of oxazoles from N-acylamino acids (**Scheme 1**). <sup>[11]</sup>

Treatment of 3-hydroxybut-2-enimidates with IBD in the presence of benzotriazole (BTA) led to the production of the rearranged 2,4,5-trisubstituted oxazoles (**Scheme 2**).<sup>[12]</sup>

Scheme 2 Oxidative rearrangement of 3-hydroxybut-2-enimidates by using IBD.

Oxazoles have just recently been synthesized by irradiating arylimidates with IBD in the presence of CsI (Scheme 3). <sup>[13]</sup>

Ar<sup>2</sup> Csl (3 equiv)  
Phl(OAc)<sub>2</sub> (3 equiv)  

$$hv$$
, PhMe, 23 °C, 24 h  
Ar<sup>1</sup> = Ph, 3-ClC<sub>6</sub>H<sub>4</sub>, 4-MeOC<sub>6</sub>H<sub>4</sub>, 1-naphthyl, etc.  
Ar<sup>2</sup> = Ph, 2-tol, 2-ClC<sub>6</sub>H<sub>4</sub>, 2-pyridyl, 4-FC<sub>6</sub>H<sub>4</sub>, etc.  
Scheme 3 Oxidative cyclization of arylimidates

## **B.** Synthesis of isoxazoles

It has been reported that nitrile oxide and alkynyliodonium salts can be used in the synthesis of 3,5- disubstituted isoxazoles (Scheme 4) <sup>[14]</sup>

$$t-Bu \sim N^{-OH} + R \longrightarrow IPh OTf OTf OHO CHCl_3, rt t-Bu \sim N^{-O}$$

R = MeOCO, *t*-BuCO, 4-Ts, etc. Scheme 4 Cycloaddition of aldoxime with alkynyliodonium salts

3,4-disubstituted isoxazoles are synthesized by the regioselective cycloaddition reaction of aldoximes with enaminones and HTIB (Scheme 5). <sup>[15]</sup>

 $R^{1} \sim N^{-}OH^{-} + R^{2} \sim NMe_{2} \qquad \frac{Phl(OH)OTs (2 equiv)}{CH_{2}Cl_{2}, rt, 1.5-24 h} \qquad R^{2}OC^{-} R^{3}$ (3 equiv)  $R^{1} = Ph, 4-tol, 4-ClC_{6}H_{4}, 2-naphthyl, BnCH_{2}, etc.$   $R^{2} = Me, \dot{k}Pr, c-Pr, EtO, Ph$ 

Scheme 5 Synthesis of Isoxazoles by regioselective cycloaddition reaction

### C. Synthesis of fused 1,2,4-triazolopyridines

Triazolopyridines were produced through the oxidation of 2-pyridylhydrazones in  $CH_2Cl_2$  by  $PhI(OAc)_2$  (Scheme 6) <sup>[16]</sup>



 $\begin{array}{l} \text{Ar = a) } C_6H_5 \text{, b) } 4\text{-}CH_3C_6H_4 \text{, c) } 4\text{-}ClC_6H_4 \text{, d) } 4\text{-}CH_3OC_6H_4 \\ \text{e) } 5\text{-}NO_2\text{-}2\text{-}furyl \text{ f) } 2\text{-}thienyl \\ \text{Scheme 6 } \text{Synthesis of } 3\text{-}aryl/heteroaryl-[1,2,4]\text{-}triazolo[4,3\text{-}a]pyridines \\ \end{array}$ 

## D. Synthesis of fused benzo-oxazolo-triazoles

IBD-mediated production of benzo[4,5]oxazolo[2,3-c][1,2,4]triazoles has been demonstrated by Demmer and co-workers (Scheme 7). <sup>[17]</sup>



Ar = a)  $C_6H_{5}$ , b) 2,6-(CH<sub>3</sub>O) $C_6H_3$ Scheme 7 IBD-mediated formation of triazoles

# E. Synthesis of fused thiazolotriazoles

Hydrazones have been subjected to oxidative cyclization in the presence of PhI(OAc)<sub>2</sub> to produce thiazolotriazoles (Scheme 8). <sup>[18]</sup>



# F. Synthesis of fused triazolopyrazines

Demmer et al. have described the synthesis of 3-Aryl/heteroaryl-[1,2,4]-triazolo[4,3-a]pyrazines from 2- chloro-3-hydrazinylpyrazine *via* IBD (**Scheme 9**).<sup>[19]</sup>



Scheme 9 Synthesis of pyrazines

#### G. C-H alkynylation of indoles and pyrroles

By utilizing the TIPS-EBX, Brand et al. reported the gold-catalyzed direct alkynylation of pyrroles and indoles (Scheme 10).<sup>[20]</sup>



Scheme 10 Alkylation of indoles and pyrroles

## H. Synthesis of Azirines

Utilizing innovative hypervalent iodine(III/V), Zhang and colleagues demonstrated the effective synthesis of 2-unsubstituted 2H-azirines by intramolecular oxidative azirination (Scheme 11).<sup>[21]</sup>



Scheme 11 Iodine (III) mediated existing and new techniques for the building of 2H-azirine skeletons.

### I. Synthesis of flavonoids

Kumar and coworkers created 2,3-dimethoxy-3-hydroxyflavanones by oxidizing several flavonol derivatives with polymer-supported HTIB (**Scheme 12**).<sup>[22]</sup>



R<sup>1</sup> = H, Cl; R<sup>2</sup> = H, OMe, Me, Cl Scheme 12 Preparation of 2,3-dimethoxy-3-hydroxyflavanones

IBD caused the synthesis of 2,3-dimethoxy-3-hydroxy-2-(1-phenyl-3-aryl-4-pyrazolyl)chromanones when it was applied to 3-hydroxy-2-(1-phenyl-3-aryl-4-pyrazolyl)chromones (**Scheme 13**).<sup>[23]</sup>



Ganguly et al. have described a one-pot procedure for producing 3',5'-diiodoflavone by iodinating the phenolic ring of o-hydroxychalcone with oxidative cyclization by using IBD (**Scheme 14**).<sup>[24]</sup>



Scheme 14 Synthesis of 3,5'-diiodoflavone and flavones by using IBD

According to a recent study by Muthukrishnan et al., the oxidation of flavanones with HTIB in the presence of ionic liquid ([bbim]<sup>+</sup>Br<sup>-</sup>) produced flavones (**Scheme 15**).<sup>[25]</sup>



Scheme 15 Synthesis of flavones in the presence of ionic liquid and HTIB

Juhasz and colleagues described how to create methyl 2-aryl-2,3-dihydrobenzofuran-3-carboxylates by employing IBD to carry out the reaction of racemic and chiral (laevorotatory) flavanone (**Scheme 16**). <sup>[26]</sup>



Scheme 16 Oxidation of chiral flavanone by using IBD-mediated method

The oxidative cyclization and rearrangement of DHA happens directly from its chalcone counterparts (Scheme 17). <sup>[27]</sup>



 $\mathbf{Ar} = C_6H_5, 4-CH_3C_6H_4, 4-OMeC_6H_4, 2-CH_3C_6H_4, 4-ClC_6H_4, 4-(CH_3)_2CHC_6H_4$ Scheme 17 Formation of isoflavone derivatives of DHA From *o*benzoyloxychalcones synthesis of isoflavones by using HTIB (Scheme 18). <sup>[28]</sup>

Recently, Zou and coworkers reported the first total synthesis of C-glycoside, 6-tert-butylpuerarin (isoflavone), by oxidative cyclization and rearrangement of C-glucosylchalcone with IBD and p- toluenesulfonic acid (**Scheme19**). <sup>[29]</sup>



# 3. CONCLUSION

The substantial group of heterocycles addressed in this chapter has recently attracted the attention of many medicinal, biological, and pharmacological chemists. The majority of heterocycles have bioactive properties that have ushered in a new era of therapeutic medicines for conditions like leprosy, cancer, hypertension, allergies, inflammatory diseases, fungal infections, and bacterial infections.

This chapter has described the useful and effective abilities of hypervalent iodine reagents in this area through the synthesis of heterocyclic compounds using organohypervalent iodine. In the coming years, there will be major research effort on the synthesis of heterocyclic compounds utilizing hypervalent iodine reagents to examine the significance of heterocycles in a variety of domains, including pharmaceutical chemistry, biochemistry, and others. It is anticipated that this coverage will lead to further substantial research in this area.

# 4. CONFLICTS OF INTEREST

The authors have to declare that there is no conflict of interest.

## 5. ACKNOWLEDGEMENTS

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